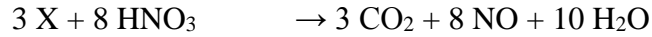


1. $\dots\dots\dots\text{H}_2 + \dots\dots\dots\text{Cl}_2 \rightarrow \dots\dots\dots\text{HCl}$
2. $\dots\dots\dots\text{C}_6\text{H}_{12}\text{O}_6 + \dots\dots\dots\text{O}_2 \rightarrow \dots\dots\text{CO}_2 + \dots\dots\text{H}_2\text{O}$
3. $\dots\dots\dots\text{BaO}_2 \rightarrow \dots\dots\text{BaO} + \dots\dots\text{O}_2$
4. $\dots\dots\text{N}_2 + \dots\dots\text{H}_2 \rightarrow \dots\dots\text{NH}_3$
5. $\dots\dots\text{Al} + \dots\dots\text{HBr} \rightarrow \dots\dots\text{AlBr}_3 + \dots\dots\text{H}_2$
6. $\dots\dots\text{Ca} + \dots\dots\text{HCl} \rightarrow \dots\dots\text{CaCl}_2 + \dots\dots\text{H}_2$
7. $\dots\dots\text{Fe}_2\text{O}_3 + \dots\dots\text{Al} \rightarrow \dots\dots\text{Al}_2\text{O}_3 + \dots\dots\text{Fe}$
8. $\dots\dots\text{KClO}_3 \rightarrow \dots\dots\text{KCl} + \dots\dots\text{O}_2$
9. $\dots\dots\text{CH}_4 + \dots\dots\text{O}_2 \rightarrow \dots\dots\text{CO}_2 + \dots\dots\text{H}_2\text{O}$
10. $\dots\dots\text{C}_2\text{H}_6 + \dots\dots\text{O}_2 \rightarrow \dots\dots\text{CO}_2 + \dots\dots\text{H}_2\text{O}$
11. $\dots\dots\text{CH}_3\text{OH} + \dots\dots\text{O}_2 \rightarrow \dots\dots\text{CO}_2 + \dots\dots\text{H}_2\text{O}$
12. $\dots\dots\text{C}_4\text{H}_{10} + \dots\dots\text{O}_2 \rightarrow \dots\dots\text{CO}_2 + \dots\dots\text{H}_2\text{O}$
13. $\dots\dots\text{C}_2\text{H}_5\text{OH} + \dots\dots\text{O}_2 \rightarrow \dots\dots\text{CO}_2 + \dots\dots\text{H}_2\text{O}$
14. $\dots\dots\text{CaCO}_3 \rightarrow \dots\dots\text{CaO} + \dots\dots\text{CO}_2$
15. $\dots\dots\text{NaOH} + \dots\dots\text{H}_2\text{SO}_4 \rightarrow \dots\dots\text{Na}_2\text{SO}_4 + \dots\dots\text{H}_2\text{O}$
16. $\dots\dots\text{NH}_3 + \dots\dots\text{O}_2 \rightarrow \dots\dots\text{NO} + \dots\dots\text{H}_2\text{O}$
17. $\dots\dots\text{H}_2\text{SO}_4 + \dots\dots\text{Ca(OH)}_2 \rightarrow \dots\dots\text{Ca(SO}_4) + \dots\dots\text{H}_2\text{O}$
18. $\dots\dots\text{H}_2\text{SO}_4 + \dots\dots\text{Al(OH)}_3 \rightarrow \dots\dots\text{Al}_2(\text{SO}_4)_3 + \dots\dots\text{H}_2\text{O}$
19. $\dots\dots\text{NaClO}_3 \rightarrow \dots\dots\text{NaCl} + \dots\dots\text{O}_2$
20. $\dots\dots\text{NH}_4\text{Cl} \rightarrow \dots\dots\text{NH}_3 + \dots\dots\text{HCl}$
21. $\dots\dots\text{NaHCO}_3 \rightarrow \dots\dots\text{Na}_2(\text{CO}_3) + \dots\dots\text{H}_2\text{O} + \dots\dots\text{CO}_2$
22. $\dots\dots\text{KOH} + \dots\dots\text{HBr} \rightarrow \dots\dots\text{KBr} + \dots\dots\text{H}_2\text{O}$
23. $\dots\dots\text{Mg} + \dots\dots\text{HBr} \rightarrow \dots\dots\text{MgBr}_2 + \dots\dots\text{H}_2$
24. $\dots\dots\text{Mg(OH)}_2 \rightarrow \dots\dots\text{Mg} + \dots\dots(\text{OH})$
25. $\dots\dots\text{NH}_3 + \dots\dots\text{H}_2\text{O} \rightarrow \dots\dots\text{NH}_4 + \dots\dots\text{OH}$
26. $\dots\dots\text{Ca(OH)}_2 \rightarrow \dots\dots\text{Ca} + \dots\dots(\text{OH})$

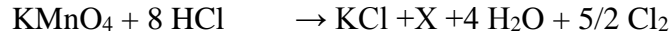
SORULAR :

1- Aşağıdaki denkleştirilmiş tepkimede X yerine hangi bileşik veya element yazılmalıdır?

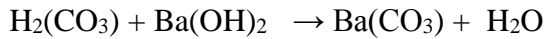
2- Aşağıdaki denkleştirilmiş tepkimede X yerine hangi bileşik veya element yazılmalıdır?



3- Aşağıdaki denkleştirilmiş tepkimede X yerine hangi bileşik veya element yazılmalıdır?



4- Aşağıdaki kimyasal tepkime denkleştirildiğinde H₂O ' nun katsayısı kaç olur?



5- Aşağıdaki kimyasal tepkime denkleştirildiğinde X ve Y yerine hangi katsayılar yazılmalıdır?

